

REPORT DOCUMENTATION PAGE				Form Approved OMB No. 0704-0188	
Public reporting burden for this collection of information is estimated to average 1 hour per response, including the time for reviewing instructions, searching existing data sources, gathering and maintaining the data needed, and completing and reviewing this collection of information. Send comments regarding this burden estimate or any other aspect of this collection of information, including suggestions for reducing this burden to Department of Defense, Washington Headquarters Services, Directorate for Information Operations and Reports (0704-0188), 1215 Jefferson Davis Highway, Suite 1204, Arlington, VA 22202-4302. Respondents should be aware that notwithstanding any other provision of law, no person shall be subject to any penalty for failing to comply with a collection of information if it does not display a currently valid OMB control number. <b>PLEASE DO NOT RETURN YOUR FORM TO THE ABOVE ADDRESS.</b>					
1. REPORT DATE (DD-MM-YYYY) 2004		2. REPORT TYPE Conference Paper POSTPRINT		3. DATES COVERED (From - To) 2001 - 2004	
4. TITLE AND SUBTITLE Substrate/oxide interface interaction in LaAlO <sub>3</sub> /Si structures				5a. CONTRACT NUMBER F29601-01-C-0241	
				5b. GRANT NUMBER	
				5c. PROGRAM ELEMENT NUMBER 62601F	
6. AUTHOR(S) T. Busani, R.A.B Devine				5d. PROJECT NUMBER 4846	
				5e. TASK NUMBER RP	
				5f. WORK UNIT NUMBER A1	
7. PERFORMING ORGANIZATION NAME(S) AND ADDRESS(ES)  Air Force Research Laboratory Space Vehicles Directorate 3550 Aberdeen Ave SE Kirtland AFB, NM 87117-5776				8. PERFORMING ORGANIZATION REPORT NUMBER AFRL-VS-PS-TP-2007-1014	
9. SPONSORING / MONITORING AGENCY NAME(S) AND ADDRESS(ES)				10. SPONSOR/MONITOR'S ACRONYM(S) AFRL/V SSE	
				11. SPONSOR/MONITOR'S REPORT NUMBER(S)	
12. DISTRIBUTION / AVAILABILITY STATEMENT Approved for public release; distribution is unlimited. (Clearance # VS03-0566)					
13. SUPPLEMENTARY NOTES Published in the Proc. Mat. Res. Soc. Symp <b>786</b> 189 (2004)					
14. ABSTRACT Amorphous lanthanum aluminate films (LaAlO <sub>3</sub> ) were deposited on Si substrates at room temperature using rf sputtering in pure Ar or an Ar/O <sub>2</sub> mixture with a stoichiometric target. The film composition was analyzed using XPS and EDX. The evolution of the material resulting from annealing at temperatures in excess of 900 °C was studied using infrared spectroscopy, XPS profiling and AFM. We obtain clear evidence for in-diffusion of Si from the substrate into the dielectric film. FTIR analysis showed only one peak centered at 747 cm <sup>-1</sup> with an FWHM of 185cm <sup>-1</sup> for as-deposited samples indicating an amorphous structure. Annealed samples showed very narrow absorptions at 483-510, 607, 695-720 and 811 cm <sup>-1</sup> . No evidence for SiO <sub>2</sub> peaks at ~1060 cm <sup>-1</sup> was observed suggesting that the LaAlO <sub>3</sub> structure tends not to reduce into a mixture of SiO <sub>2</sub> and a silicide. Short time annealing at 1000 °C results in a broad band at 905 cm <sup>-1</sup> which can be interpreted in terms of the formation of a layer rich in Si-O-La bonds. Nitridation of the substrate before oxide deposition and annealing slows the degradation process but does not suppress it. X-ray diffraction analysis of the annealed films indicates a very oriented crystalline structure, yet unidentified, whose direction depends upon the orientation of the Si substrate. The dielectric constant in both annealed and as-deposited films was measured to be less than 14 and the leakage current density was very low. Some mobile charge was detected. This dielectric constant is substantially less than the value ~25 anticipated from bulk, single crystal measurements.					
15. SUBJECT TERMS LaAlO <sub>3</sub> , Amorphous Films, Dielectric Film					
16. SECURITY CLASSIFICATION OF:			17. LIMITATION OF ABSTRACT  Unlimited	18. NUMBER OF PAGES  7	19a. NAME OF RESPONSIBLE PERSON: Capt John Boyd
a. REPORT Unclassified	b. ABSTRACT Unclassified	c. THIS PAGE Unclassified			19b. TELEPHONE NUMBER (include area code) 505-853-3157

### Substrate/oxide interface interaction in $\text{LaAlO}_3/\text{Si}$ structures

T. Busani<sup>1</sup> and R.A.B. Devine

AFRL-VSSE, Kirtland Air Force Base, Albuquerque, NM 87117

<sup>1</sup>also with Université Joseph Fourier, Grenoble, France

#### ABSTRACT

Amorphous lanthanum aluminate films ( $\text{LaAlO}_3$ ) were deposited on Si substrates at room temperature using rf sputtering in pure Ar or an  $\text{Ar}/\text{O}_2$  mixture with a stoichiometric target. The film composition was analyzed using XPS and EDX. The evolution of the material resulting from annealing at temperatures in excess of 900 °C was studied using infrared spectroscopy, XPS profiling and AFM. We obtain clear evidence for in-diffusion of Si from the substrate into the dielectric film. FTIR analysis showed only one peak centered at 747  $\text{cm}^{-1}$  with an FWHM of 185  $\text{cm}^{-1}$  for as-deposited samples indicating an amorphous structure. Annealed samples showed very narrow absorptions at 483-510, 607, 695-720 and 811  $\text{cm}^{-1}$ . No evidence for  $\text{SiO}_2$  peaks at ~1060  $\text{cm}^{-1}$  was observed suggesting that the  $\text{LaAlO}_3$  structure tends not to reduce into a mixture of  $\text{SiO}_2$  and a silicide. Short time annealing at 1000 °C results in a broad band at 905  $\text{cm}^{-1}$  which can be interpreted in terms of the formation of a layer rich in Si-O-La bonds. Nitridation of the substrate before oxide deposition and annealing slows the degradation process but does not suppress it. X-ray diffraction analysis of the annealed films indicates a very oriented crystalline structure, yet unidentified, whose direction depends upon the orientation of the Si substrate. The dielectric constant in both annealed and as-deposited films was measured to be less than 14 and the leakage current density was very low. Some mobile charge was detected. This dielectric constant is substantially less than the value ~ 25 anticipated from bulk, single crystal measurements.

#### INTRODUCTION

The dramatic expansion in technology and the communications market, including the one associated with high performance microprocessors as well as low-static power applications such as wireless systems [1], has driven industry to require greater integrated circuit functionality and performance at lower costs. This means an increase in circuit density for each Si wafer [2]. Obviously, device channel length and the  $\text{SiO}_2$  gate dielectric thickness must decrease. However,  $\text{SiO}_2$  layers thinner than ~ 1.3 nm have excessive leakage current ( $> 1 \text{ A}/\text{cm}^2$ ) simply due to direct tunneling through the oxide, and this is far too large for future circuits [3]. Unfortunately, the Semiconductor Industry Association road map indicates that a sub-1 nm effective oxide thickness (EOT) will be required for complementary metal-oxide-semiconductor (CMOS) devices with 50 nm gate widths so a dilemma exists.

An acknowledged solution (in principal) to the leakage current issue is to maintain the gate capacitance by increasing the dielectric constant ( $k$ ) of the oxide and simultaneously thicken it. Though it is unlikely that the alternative oxide will have the excellent leakage properties of thermally grown  $\text{SiO}_2$ , a compromise may exist. Many materials acknowledged as "high- $k$ ", have been proposed in the last few years. They must satisfy a variety of criteria, such as thermodynamic stability in contact with the Si channel, potential barriers to both electron and hole injection from the substrate/inversion layer, low leakage, etc. Amongst the most commonly

studied materials were  $\text{Al}_2\text{O}_3$ ,  $\text{Ta}_2\text{O}_5$  and  $\text{SrTiO}_3$ , because they have a dielectric constant running from 9 to 80 and a band gap from 3.2 to 8.8 [4-5]. Interfacial reaction, resulting in the formation of  $\text{SiO}_2$  or some low-k layer, has been observed in the case of  $\text{Ta}_2\text{O}_5$  on Si [6] and in a  $\text{Ta}_2\text{O}_5$  sandwich [7]. Furthermore, the barrier height to carrier injection is estimated to be rather small ( $< 0.5$  eV).  $\text{Al}_2\text{O}_3$  is potentially an attractive short term solution because of its relatively low dielectric constant ( $\sim 9$ ) but studies of  $\text{Al}_2\text{O}_3$  deposited on Si(100) at temperatures below  $400^\circ\text{C}$  [8] show evidence of an Aluminum silicate phase formed at the interface with Si. Because of this, an alternative gate dielectric needs to be found for a long-term industry solution.

$\text{LaAlO}_3$  has been investigated as an alternative dielectric since it is known that the La-O bonds in the metal oxide are strongly polarizable resulting in an increased dielectric constant. The crystalline phase of  $\text{LaAlO}_3$  has  $k \sim 28$  and a band gap of 5.6 eV [5]. Many studies have demonstrated that it is possible to deposit  $\text{LaAlO}_3$  layers of several nm onto Si with good electrical properties using different techniques [9-10]. The film is usually, however, amorphous. All these depositions were carried out at relatively high temperature ( $\geq 600^\circ\text{C}$ ). Some of the films were recrystallized ex-situ.

In the present work we analyze in detail the effects of temperature on films deposited initially at room temperature. We demonstrate that it is not possible to obtain a desired crystalline phase for  $\text{LaAlO}_3$  from this starting point.

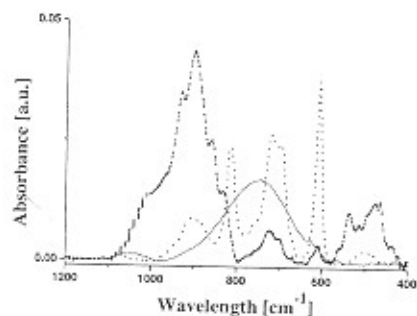
## EXPERIMENTS

p-type Si(100) 3-5 ohm/cm resistivity wafers were first rinsed in pure HF acid then blown dry. Thin  $(\text{La}_x\text{Al}_{1-x})_2\text{O}_3$  were deposited at room temperature by sputtering from a polycrystalline  $\text{LaAlO}_3$  target using a 13.56 MHz radio frequency (rf) sputtering source (TORUS 2) with an  $\text{Ar}/\text{O}_2$  gas mixture. The chamber pressure was 10 mTorr and the gas flow rates were set respectively at 75 sccm and 5 sccm. To optimise the stoichiometry of the deposited films the typical rf power was 60 W [9], calculations of the sputtering yield indicated that this power was appropriate for equal sputtering rates of the La and Al. Under these conditions  $(\text{La}_x\text{Al}_{1-x})_2\text{O}_3$  films were obtained with  $x=0.6$  as determined by x-ray photoelectron spectroscopy (XPS) profiling and electron microscopy spectroscopy (EDS) measurements. The rate of deposition of the film was typically 1.4 nm/min, typical film thicknesses studied were  $\sim 100$  nm. A second set of samples was prepared in the same conditions, but initially the Si surface was exposed to an  $\text{N}_2$  plasma excited at 2.45 GHz. Such exposure is known to generate an ultra-thin  $\text{Si}_3\text{N}_4$  layer, probably less than a mono-layer. The  $(\text{La}_x\text{Al}_{1-x})_2\text{O}_3$  thickness and refractive index ( $n$ ) for all films was determined using single wavelength (632.8 nm) ellipsometry, the thickness was confirmed by comparison of the Fourier Transform Infrared (FTIR) absorption and comparison with preceding work [9]. Surface analysis and roughness was carried out using an Autoprobe CP Atomic Force Microscopy (AFM) in a non-contact mode. The force used was 13.6 nN with a scanning speed of 3 Hz. Capacitance-Voltage (C-V) curves were measured at 100 KHz using a Keithley 590 system after deposition of Al dots of  $0.0078 \text{ cm}^2$  on the dielectric surface. The back face of the wafer was metallized to ensure good electrical contact. Prior to electrode deposition the wafers were cleaved and some samples were subjected to rapid thermal annealing between  $900$  to  $1000^\circ\text{C}$  for 1 to 5 min. in  $\text{N}_2$  or  $\text{N}_2\text{H}_2$  atmospheres in an attempt to minimize La and Al interactions with the silicon substrate. Finally X-ray diffraction intensity data was obtained in a  $0.02^\circ$  step interval in the  $2\theta$  range between  $10$  and  $80^\circ$  with a measurement time of 1.2 sec. A  $\text{Cu}\alpha_1$  source was used.

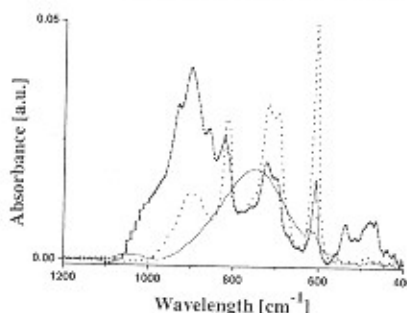
## RESULTS AND DISCUSSION

$\text{LaAlO}_3$  has a ideal perovskite cubic structure  $\text{Pn}3\text{m}$  which presents three dominant IR active modes:  $\nu_1$  corresponds to stretching of O atoms along the Al-O-Al line,  $\nu_2$  is perpendicular to this line, and  $\nu_3$  is the La translation respect to Al and O atoms. This crystal is typically presented as a rhombohedral deformation with two arrangements for the fundamental cell (space group  $\text{R}\bar{3}\text{m}$ ). The octahedral  $\text{AlO}_6$  are quite regular with an Al-O length of 0.190 nm, the bond length of La-O is  $\sim 0.25$  nm [11]. We observe that the bond force constants for La-O and La-Al are weak compared to Al-O in the  $\text{AlO}_6$  octahedra. This helps us to separate the vibrational modes into *internal*, when we considered the movements of Al or O in the  $\text{AlO}_6$  cell, and *external*, related to the La movement respect to  $\text{AlO}_6$  octahedra. Since the crystal symmetry is lower, some IR inactive modes become active and new vibrations may appear in the spectra. An example is the  $\nu_1$  vibration which is predicted to be split into two peaks depending upon crystal deformation. Saine et al. [11] give the relation of the  $\text{AlO}_6$  modes among cubic, rhombohedral and isolated structures.

In Figure 1a we show typical absorbance IR spectra of the room temperature deposited films.



**Figure 1a.** Normal incidence absorbance IR spectra of  $(\text{La}_x\text{Al}_{1-x})_2\text{O}_3$  deposited on a Si wafer. The solid line refers to as-deposited oxide, the dotted line is after annealing at 950 °C in an  $\text{N}_2$  atmosphere for 1 min. and the square dashed line is after annealing at 1000 °C for 1 min.



**Figure 1b.** Normal incidence absorbance IR spectra of  $(\text{La}_x\text{Al}_{1-x})_2\text{O}_3$  deposited on a previously nitrided Si wafer. The solid line refers to as-deposited oxide, the dotted line is for the sample after annealing at 950 °C in  $\text{N}_2$  atmosphere for 1 min. and the square dashed line is for a sample annealed at 1000 °C for 1 min.

The solid line is representative of the as-deposited oxide and shows one intense peak at  $747\text{ cm}^{-1}$  and a very small peak around  $1050\text{ cm}^{-1}$ . The latter is due to the formation of an initial  $\text{SiO}_2$  layer during ignition of the Ar and  $\text{O}_2$  plasma. The evidence of an amorphous structure is confirmed by the full width at half maximum (FWHM) of  $185\text{ cm}^{-1}$  of the main peak. One can very crudely approximate:

$$\nu = \frac{1}{\lambda} = \frac{1}{2\pi c} \sqrt{\frac{f}{\mu}} \quad (1)$$

where  $\nu$  is the vibration mode frequency in a  $B-O$  molecule ( $B$  being Al or La) expressed in  $\text{cm}^{-1}$ ,  $\mu = (m_B m_O) / (m_B + m_O)$  is the reduced mass,  $m_O$  and  $m_B$  are the atomic weight in Kg of cation  $B$  and anion  $O$ ,  $f = 17/r^2$  is the stretching force constant and  $r$  is the  $B-O$  length. We can estimate the theoretical frequency in the pure, binary glasses  $B_2O_3$  using the bond lengths for La-O and Al-O [12-13]. We find that the stretching modes are respectively at 355 and 778  $\text{cm}^{-1}$ . From eq. (1), the wave number of the vibration is inversely proportional to the value of the reduced mass. In consequence, we can conclude that the presence of La in  $\text{LaAlO}_3$  glass probably has the effect of moving the IR spectra to lower frequencies than that for pure  $\text{Al}_2\text{O}_3$  which is consistent with the result for the amorphous phase shown in Figure 1a.

The dotted line in Figure 1a shows a typical IR absorbance spectra for a sample annealed at 950 °C for 1 min. We observe similar spectra in samples where the anneal was carried out at 950 °C for 15 min and 950 °C for 4 min, in an  $\text{N}_2$  or  $\text{N}_2\text{H}_2$  atmosphere. First we note that the intensity of  $\text{SiO}_2$  small peak does not increase, suggesting that little or no oxygen from the  $(\text{La}_x\text{Al}_{1-x})_2\text{O}_3$  diffuses and reacts at the Si substrate interface. There are 4 very narrow peaks at 483-510, 607, 695-720 and 811  $\text{cm}^{-1}$  indicating a polycrystalline or local crystalline structure. The 695-720  $\text{cm}^{-1}$  peak is the  $\nu_1$  vibration of the octahedral  $\text{AlO}_6$  and the mode at 510-483  $\text{cm}^{-1}$  could be attributed to  $\nu_2$  mode. According to Saine [11] in the rhombahedral structure we should see another peak at 565  $\text{cm}^{-1}$  related to the  $\nu_1$  mode in a cubic cell while a peak at 511  $\text{cm}^{-1}$  is related to the angular deformation of O-Al-O due to phase movements between two octahedra. This vibration is impossible in the  $\text{AlO}_6$  isolated structure. Abrashev et al. [15] have calculated the transverse optic/longitudinal optic (TO/LO = 481/505  $\text{cm}^{-1}$ ) peak positions for the  $\nu_2$  mode. Normally, we do not expect to observe LO peaks, since our IR spectra were measured at normal incidence. However, as shown in Table I, our annealed samples have a significant roughness of several tens of nm, which suggests that the IR beam does not see a flat surface and with such roughness the beam incidence will be non-normal for parts of the surface. Under these conditions the Berreman effect [16] will be active and the LO mode may be excited. Unfortunately  $\nu_3$  is too low in wave number to be visible by our instrument and this does not help to confirm the presence of  $\text{LaAlO}_3$  crystal structure.

Sample description	Max height between adjacent peaks [nm]	Average roughness [nm]
$(\text{La}_x\text{Al}_{1-x})_2\text{O}_3$ as-deposited / Si substrate	2.25 / 2	<1
$(\text{La}_x\text{Al}_{1-x})_2\text{O}_3$ ann 950 °C 1 min / Si substrate	60-70 / 60	18.18
$(\text{La}_x\text{Al}_{1-x})_2\text{O}_3 + \text{Si}_3\text{N}_4$ ann 950 °C 1 min / Si substrate	37 / 3.6	11.27

Table I. AFM surface analysis data for different samples. The words "Si substrate" indicate the AFM is carried out on the Si after the oxide layer has been removed.

There are other arguments to suggest that these peaks are not related to a  $\text{LaAlO}_3$  single phase. If one plots the intensity of the peaks as a function of film thickness, the resultant curve does not go through zero, which it should if one deals with peaks emanating from the same vibration. The XRD data presented in Figure 2 demonstrates that the oxide is in a crystalline phase, but that it is not that of idealized  $\text{LaAlO}_3$ .

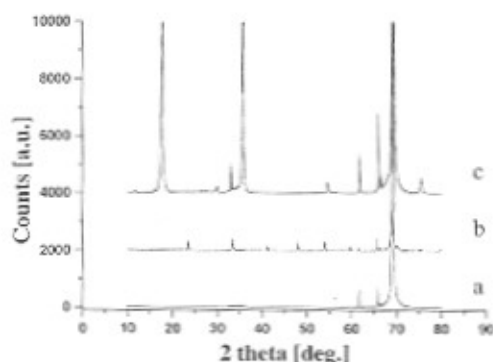


Figure 2. XRD of

- a) Si (100) substrate,
- b)  $\text{LaAlO}_3$  polycrystalline powder on Si (100) and
- c)  $(\text{La}_{0.8}\text{Al}_{0.2})_2\text{O}_3$  deposited on Si and annealed at 950 °C for 1 min.

We also identify a broad peak at  $\sim 900 \text{ cm}^{-1}$  that is the beginning of the interfacial reaction between Si and La. This interaction has been reported [14] as a vibration of the RE-O-Si bond (RE = rare-earth) in different oxides. It is evident that increasing the annealing temperature to 1000 °C, even if for 1 min, results in this peak becoming predominant and the others decrease considerably. The peak at  $808 \text{ cm}^{-1}$  is less intense, even if still important, and we identify this with the presence of an aluminosilicate structure [17]. We can conclude that Si starts to form more La-O-Si but also Si-Al-La bonds containing oxygens. We have tried to eliminate this effect using an  $\text{Si}_3\text{N}_4$  protecting layer as described previously, the results are shown in Figure 1b. When compared with Figure 1a, it appears that the silicon nitride layer does not suppress the annealing phenomenon, but it does tend to limit it.

The AFM (Table I) and XPS (Figure 3) data further help in the understanding of interfacial degradation and reaction in the  $(\text{La}_{0.8}\text{Al}_{0.2})_2\text{O}_3/\text{Si}$  system. The measurements were carried out first on the annealed and unannealed samples then the oxide was removed and Si surfaces were analysed. The roughness of the oxide surface is really significant in all samples that

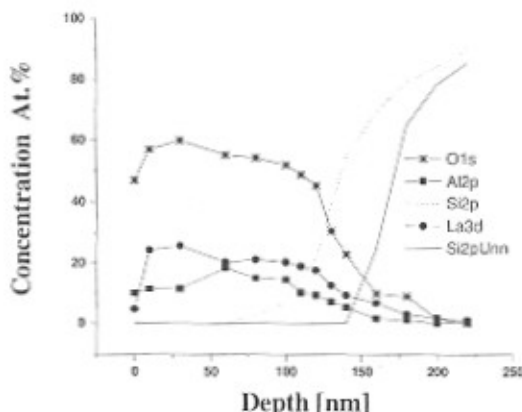


Figure 3. XPS data of  $(\text{La}_{0.8}\text{Al}_{0.2})_2\text{O}_3$  deposited on Si and annealed at 950 °C for 1 min. The profile for "Si2pUnn" refers to as-deposited samples. The Si signal appears before the Si substrate is encountered. For the amorphous (as-deposited) case (solid line), there is no evidence for Si in the oxide layer.

received a thermal treatment, but less important on that where  $\text{Si}_3\text{N}_4$  layer was grown prior to deposition. These observations are interpreted as indicating that Si diffused in the oxide layer during the high temperature anneals leaving a "roughened" structure. The XPS profile data is shown in Figure 3. Though there is always some imprecision in XPS about the exact location of the dielectric/Si interface, we clearly evidence that in an annealed sample, Si appears to be present well into the region we call the "dielectric". In fact, an anneal at 950 °C for 1 minute appears to suffice for Si to penetrate ~ 40 nm. Note that this exceeds by a factor ~ 2 the average roughness of this sample so the Si signal does not arise from the substrate, only from the film. All of these observations are consistent with the appearance of an La-O-Si associated vibrational mode in the infrared absorption spectrum as a direct result of annealing. Finally, C(V) and current density electric field (I-V) measurements were carried out to test the electrical quality of the interface and the oxide. For the amorphous, as-deposited films we find  $k$  lower than 14 and the leakage current density ~  $10^{-8}$  A/cm<sup>2</sup> for an electric field ~ 1 MV cm<sup>-1</sup>. These results are consistent with earlier findings [9]. A hysteresis effect is also visible and this does not reduce when moderated annealing, say 400 °C for 30 min in a inert atmosphere, is employed even though the flat band voltage moves from -10 V to -1.5 V.

In the light of these experiments, we are not surprised that is hard to obtain a high  $k$  oxide. The crystal phase of  $\text{LaAlO}_3$  fails to appear before significant interaction with the underlying Si has occurred. In order to employ  $\text{LaAlO}_3$  in a high  $k$  application ( $k \sim 25$ ) it is necessary to find a means of depositing densified or crystallized material and to avoid significant post-deposition annealing at high temperatures.

#### REFERENCES

1. <http://public.itrs.net>
2. G. Baccarani, M.R. Wordeman, and R.H. Dennard, IEEE Trans. Electron Devices **31**, 452 (1984)
3. S.M. Lo, D.A. Buchanan, Y. Taur, and W. Wang, IEEE Electr. Dev. Lett. **18**, 209 (1997)
4. G.D. Wilk, R.M. Wallace, J.M. Antony, J. Appl. Phys. **89**, 5243 (2001)
5. P.W. Peacock and J. Roberts, J. Appl. Phys. **92**, 4712 (2002)
6. G.B. Alers, D.J. Werder, Y. Chabal, H.C. Lu, E.P. Gusev, E. Garfunkel, T. Gustafsson, and R. S. Urdhal, Appl. Phys. Lett. **73**, 1517 (1998)
7. P.K. Roy and I.C. Kizilyalli, Appl. Phys. Lett. **72**, 2835 (1998).
8. T.M. Klein, D. Niu, W.S. Epling, W. Li, D.M. Maher, C.C. Hobbs, R.I. Hedge, I.J.R. Baumvol, and G.N. Parsons, Appl. Phys. Lett. **75**, 4001 (1999)
9. R. A. B. Devine, J. Appl. Phys. **93**, 9938 (2003)
10. Byung-Eun Park, H. Ishiwara, Appl. Phys. Lett. **82**, 1197 (2003)
11. M.C. Saine, E. Husson, and H. Brusset, Spectrochimica Acta **73A**, 985 (1981)
12. E.F. Archibong, and A. St-Amant, J. Phys. Chem. A **103**, 1109 (1999)
13. R. El-Mallawany, A. Abdel-Kader, M. El-Hawary, and N. El-Khoshkhany, Eur. Phys. J. AP **19**, 165 (2002)
14. H. Ono, Appl. Phys. Lett. **78**, 1832 (2001)
15. Abrashev, M.V., Litvinchuk, A.P., Iliev, M.N., Meng, R.L., Popov, V.N., Ivanov, V.G., Chakalov, R.A.; Thomsen, C., Phys. Rev. B **59**, 4146 (1999)
16. D.W. Berreman, Phys. Rev. **130**, 2193 (1963)
17. W. Mozgawa, and M. Sitarz, J. Mol. Struc. **614**, 273 (2002)